M.Phil. –Chemistry [Semester – II] Paper III – [Elective] INORGANIC CHEMISTRY {50 MARKS}

- 1. Co- ordination Chemistry { 25 marks }
- 2. Chemistry of Lanthanides and actinides sperations. Spectral and magnetic properties – Use of lanthanide compounds as shift reagents – organo metallic chemistry – organometallic reagents in organic synthesis and in homogeneous catalytic reaction (hydrogenation, hydroformylation, isomerisation and polymerisation) p-acidmetal complexes – activation of small molecules by co-ordinations.

References:

- 1. Introduction to Ligand field theory
 - -By Carl. J. Ballhalusen
- 2. Co-ordination compounds

-By S.F.A.Kettle

- 3. Modern Co-ordination Chemistry
 - -By F. Lewils and R.G.Wilkins

2. Group Theory {25 marks}

- 1. Hybrid orbitals and molecular orbitals for AB_n Type molecules Transformation properties of atomic orbitals – Hybridization scheme for 6 orbitals – Hybridization schemes for 'pi' bonding – Molecular orbitals theory for AB_n – Type molecules – relationship of the molecular orbital and the hybridization treatments – molecular orbitals of metal sandwhitch compounds.
- Molecular vibrations The symmetry of Normal vibrations Determining the symmetry types of the normal modes – Contribution of particular internal co-ordinates to Normal modes – calculation of force constants – selection rules for fundamental vibrational transitions.

References:

1. Group theory in chemistry

-By V.Ramakrishnan & M.S.Gopinathan

2. Physical methods in Inorganic chemistry

-By R.S.Gargo

3. Symmetry in Chemistry

-By Jaffe & orelaan